

Online and Blended Learning: How the New Approaches to Learning Work in the World of STEM

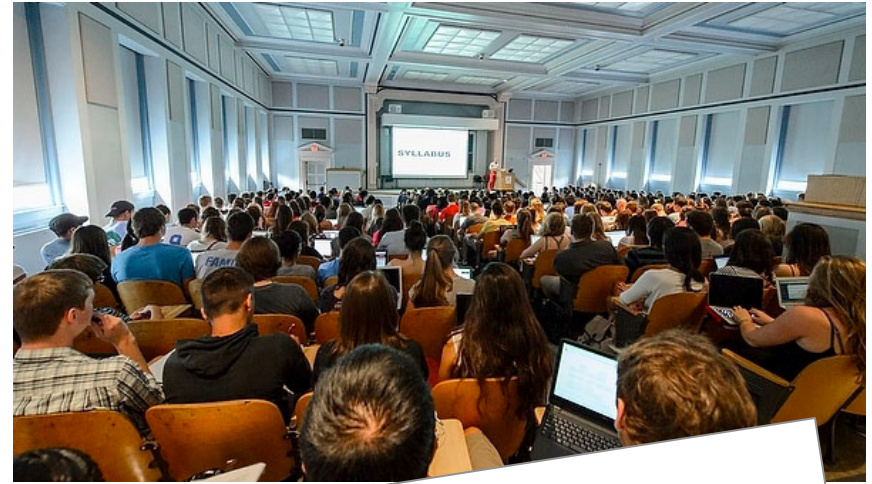
James Caras & Mats Selen

Trends

Problems to solve:

- Cost savings needed
- STEM enrollment increase


Result: Shift towards larger classrooms in foundational STEM courses



INSIDE HIGHER ED TEACHING AND LEARNING

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The STEM Enrollment Boom

 Bioengineering students at SUNY Binghamton

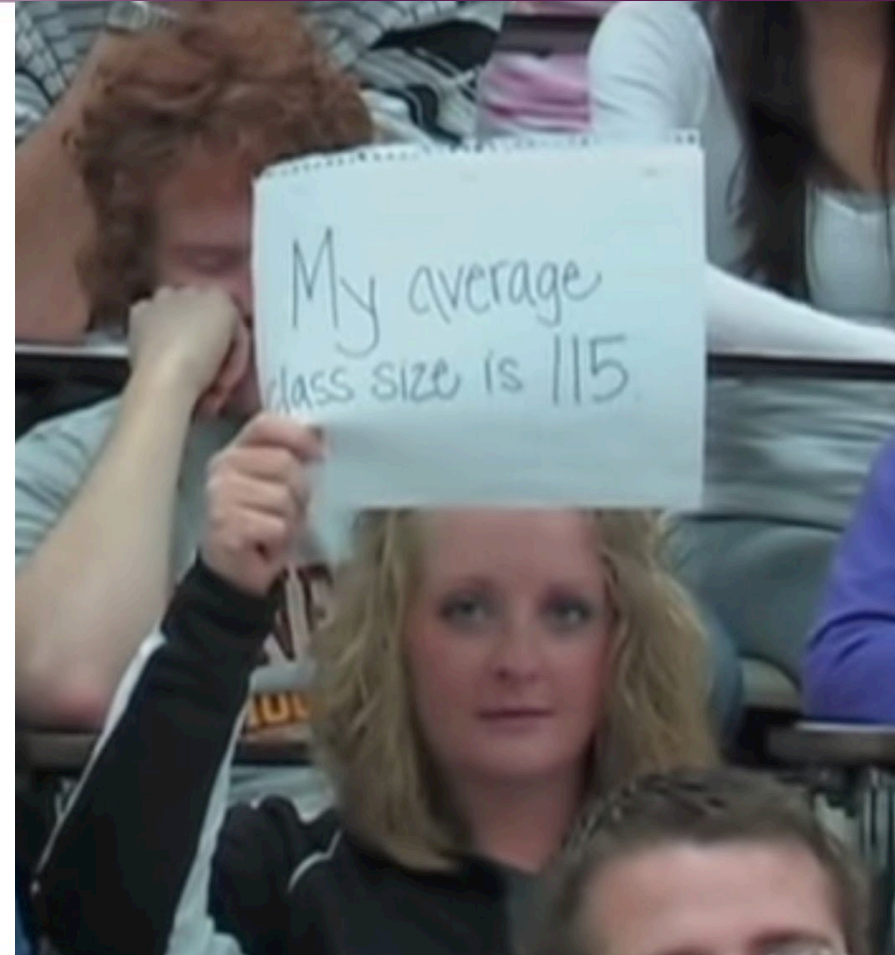
Since the recession, undergraduate enrollments have gone up dramatically, but primarily in engineering and biology and not at expense of humanities and social sciences, study finds.

Size Matters: Large Class Impact

- Decreased student engagement
- Decreased outcomes
- Decreased student satisfaction / course evaluations

Lower Engagement in Large Classes

- Lower attendance
- Lower classmate interaction and support
- Lower student preparedness for class
- Lower class participation



Attendance and Outcomes

- Class size matters: Smaller the class, **the less** absenteeism
- More absenteeism in foundational courses
- Students with strong attendance achieve one letter grade higher performance
- *How do we get students to go to large foundational classes?*

Class Size and Outcomes

- Recent studies analyzing class size and outcomes in higher education indicate larger class sizes have:
 - Little to no impact on fact retention
 - Negative impact on attainment of **higher-order thinking skills**:
 - Problem-solving
 - Written and collaborative expression
 - Critical thinking
- } Problem for STEM?
- *Why? Deep engagement required*

Outside Class Problem-Solving

- **Problem to solve:** shift from hand-graded homework to ungraded “suggested problem sets” *which students don't do*
- **Solution:** Hybrid course with online homework component
 - Automatically graded
 - Built in feedback, instruction, and tutoring

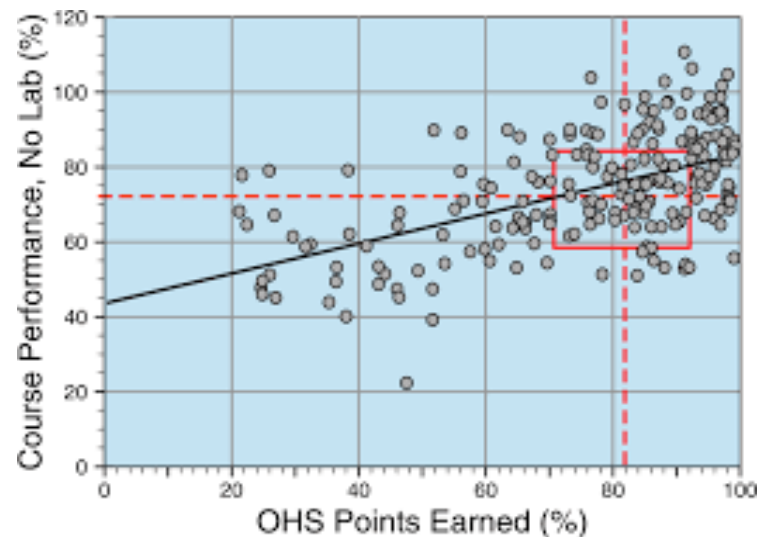
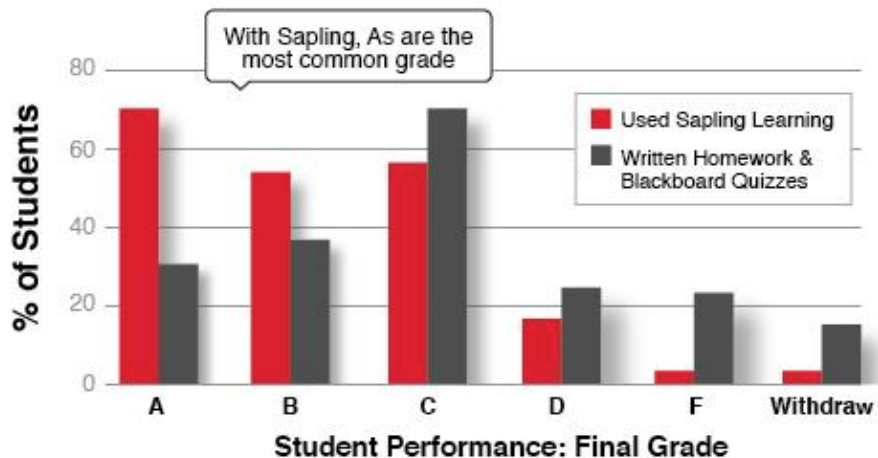
The screenshot shows a web-based question interface from Sapling Learning. The question asks to classify four substances based on their molecular diagrams. The substances are represented by four circular diagrams:

- Diagram 1: A collection of individual, non-bonded atoms of two different colors (red and blue).
- Diagram 2: A collection of identical diatomic molecules, each consisting of two blue atoms bonded together.
- Diagram 3: A collection of identical diatomic molecules, each consisting of one red atom and one blue atom bonded together.
- Diagram 4: A mixture of two different diatomic molecules: one with two blue atoms bonded together, and another with one red and one blue atom bonded together.

The interface includes a header for 'Question 1', the Sapling Learning logo, and a 'Map' icon. Below the question text, there are four empty boxes for classification, labeled 'Element', 'Compound', 'Homogeneous Mixture', and 'Heterogeneous Mixture'. At the bottom, there are navigation buttons: 'Hint', 'Previous', 'Give Up & View Solution', 'Check Answer', 'Next', and 'Exit'.

Online Homework Outcomes

Student Performance



- Numerous studies show ~1 letter grade improvement returning to graded online homework

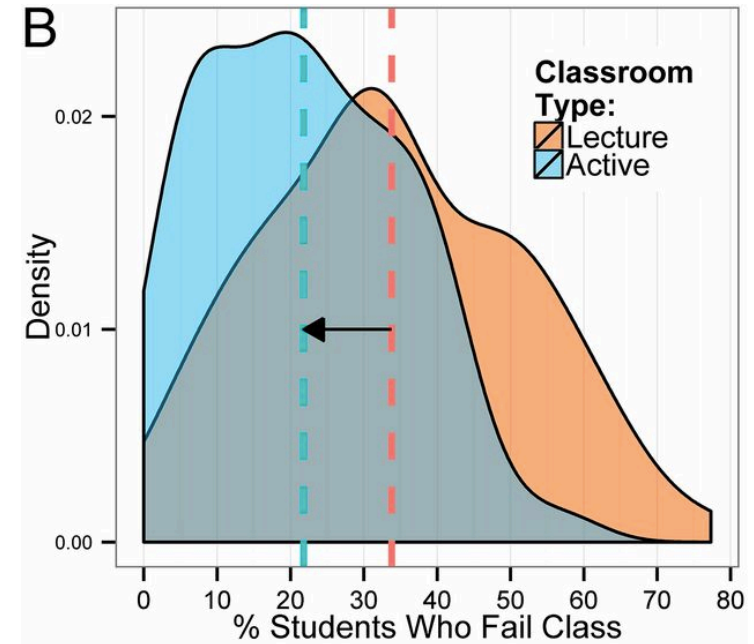
Valuable Class Time: Enable Active Learning

Problems to solve:

- Class attendance most valuable to student success: *how do we make better use of classroom time?*
- Students walking into lectures with little exposure to covered topics makes active learning difficult – *how do we free up class time?*

Solutions:

- Pre-lecture and pre-lab activities
- Active in-class engagement with student response system



Active learning increases student performance in science, engineering, and mathematics

Scott Freeman^{a,1}, Sarah L. Eddy^a, Miles McDonough^a, Michelle K. Smith^b, Nnadozie Okoroafor^a, Hannah Jordt^a, and Mary Pat Wenderoth^a

^aDepartment of Biology, University of Washington, Seattle, WA 98195; and ^bSchool of Biology and Ecology, University of Maine, Orono, ME 04469

Edited* by Bruce Alberts, University of California, San Francisco, CA, and approved April 15, 2014 (received for review October 8, 2013)

■ ■ ■

of controls over student quality and instructor identity. This is the largest and most comprehensive metaanalysis of undergraduate STEM education published to date. **The results raise questions about the continued use of traditional lecturing as a control in research studies, and support active learning as the preferred, empirically validated teaching practice in regular classrooms.**

Feedback, Data, and Analytics

- Key benefit of online learning platforms is *creating a complete picture of learning* with analytics
- Enabling learning science: data driven instruction and course design
- Challenge is:
 - Actionable data for the student and instructor
 - Supporting learning outcomes as defined by the institution

360° Focus on Hybrid

- Required:
 - Tools: hardware, software
 - **Training: Dedication to faculty Professional Development**

Example: Intro Physics at Illinois



Mechanics (calc) :1254

Electricity & Magnetism (calc) : 831

Thermo & Stat. Mech. (calc) : 692

Quantum Physics (calc) : 687

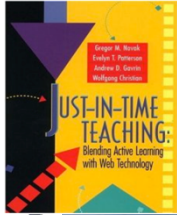
Mechanics & Heat (alg) : 331

E&M and Modern (alg) : 358

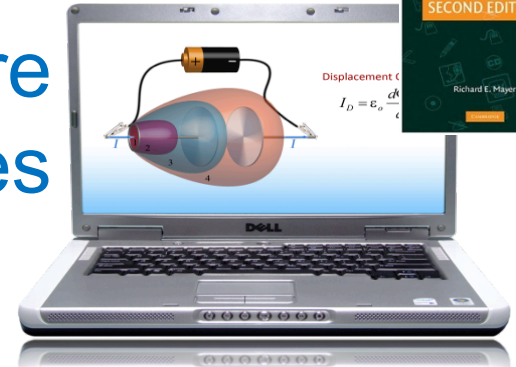
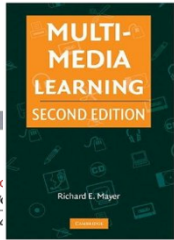
Big. Complex. Many moving parts.

Great platform for education research

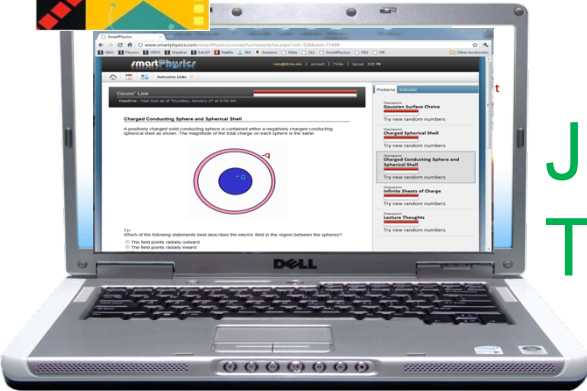
“Flip” to optimize class-time



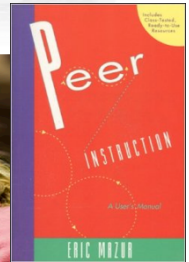
Pre
Lectures



Just in Time
Teaching



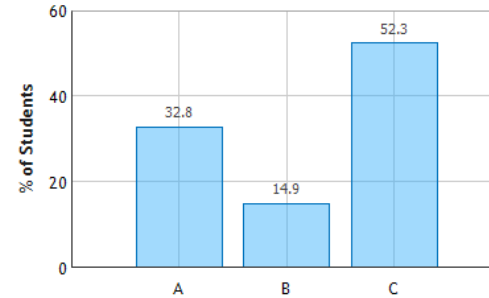
Peer
Instruction



Just in Time Teaching Example



Ice Melting in Cup of Water: Question 1 (N = 1065)



An ice cube is floating in a glass filled with water.
When the ice melts, the level of the water will:

- A) Go up, causing water to spill out of the glass
- B) Go down
- C) Stay unchanged

Student explanations for “it spills over” ...

	water in ice must add somewhere to water in glass
mail	The water level will rise. I learned that from Al Gore in "An Inconvenient Truth"
email	since the ice completely melts the volume of ice not in the water will exceed the volume of the glass and surface tension will unlikely hold the excess in place
	dope
mail	because the ice floats with some of its volume above the water, when it melts it will rise the level of the water
	the melted water has a higher density as ice and will flow into the cup to cause the overflow
email	meep!
	The same percentage of the ice's volume is always above the water, which means when the ice cube gets smaller, the volume that is above gets smaller, and the extra volume makes the water spill.
mail	1
email	The density will increase.
mail	same volume but there is water floating above the cup in the form of ice
mail	well the volumes gotta go somewhere am I right???

Lecture Slide:



An ice cube is floating in a glass filled with water.
When the ice melts, the level of the water will:

A) Go up, causing water to spill out of the glass

The water level will rise. I learned that from Al Gore in "An Inconvenient Truth"

B) Go down

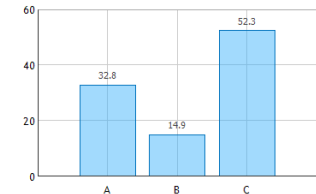
Volume of ice is greater than volume of water

C) Stay unchanged

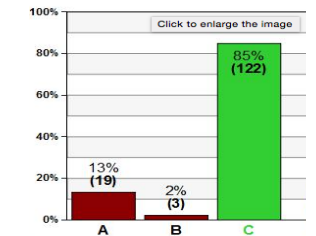
The melted water has exactly the same mass as the ice cube



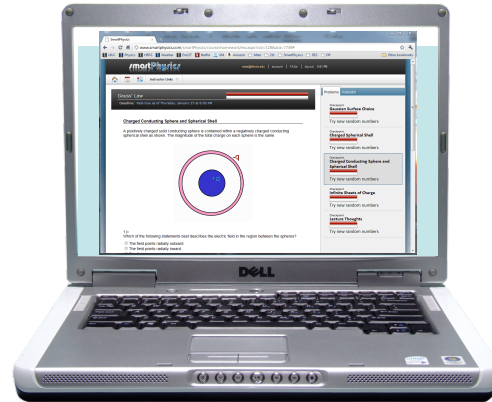
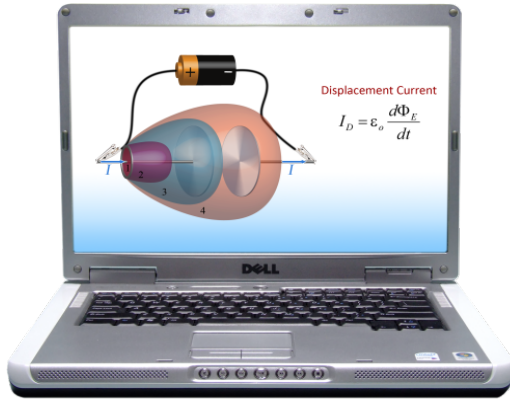
Before Class



During Class



How does this approach impact our students?

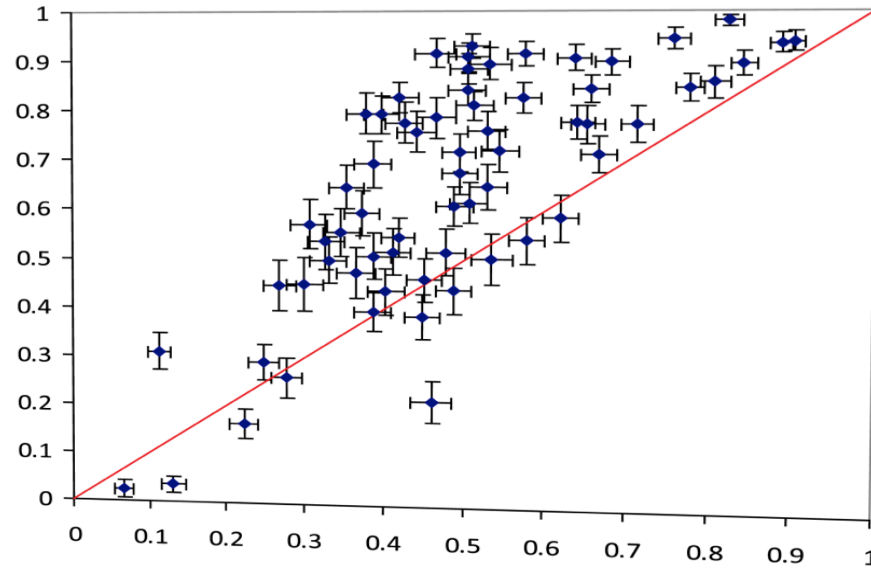


Am. J. Phys. 78, 755-759, 2010

Phys. Rev. ST Phys. Educ. Res. 6, 1-5, 2010

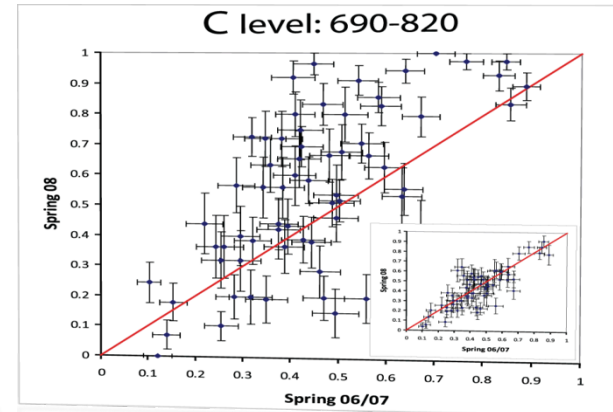
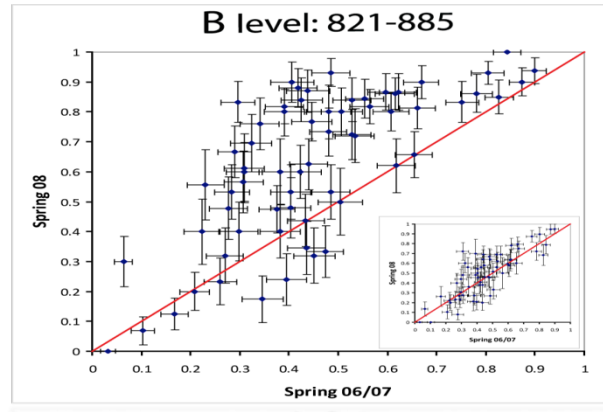
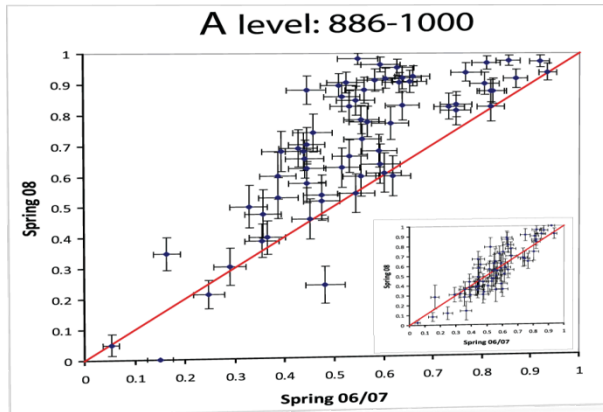
Understanding of E&M Concepts

After

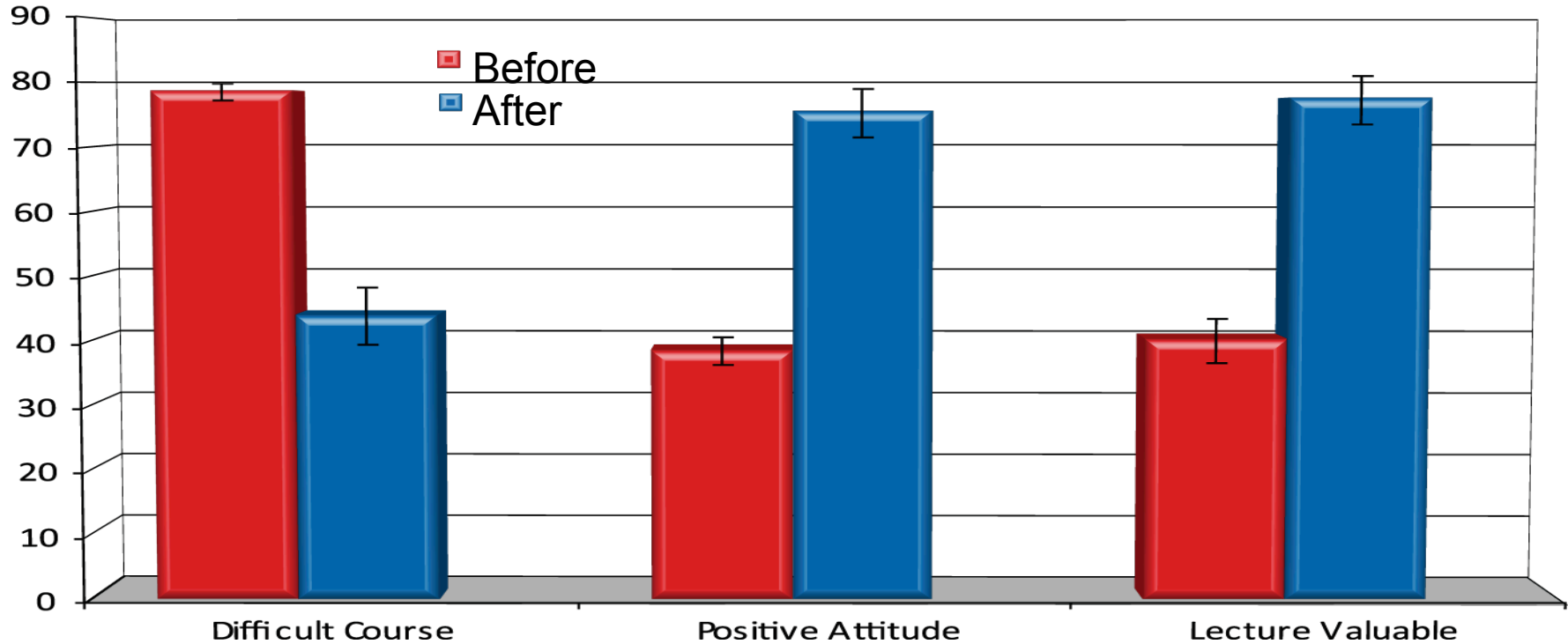


Before

Improvement seen for all students



Student Perception of the Course



Lab Reforms:

Encourage critical thinking & creativity



Why are we doing this now ?

Research

Measuring the impact of an instructional laboratory on the learning of introductory physics

American Journal of Physics **83**, 972 (2015);



Carl Wieman

Department of Physics, Stanford University, Stanford, California 94305 and Graduate School of Education, Stanford University, Stanford, California 94305

N. G. Holmes

Department of Physics, Stanford University, Stanford, California 94305

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The average benefit on lab-related questions for students who took the lab, in both courses, was within 0.6% of the score of students who did not, ...



Design and Reflection Help Students Develop Scientific Abilities: Learning in Introductory Physics Laboratories

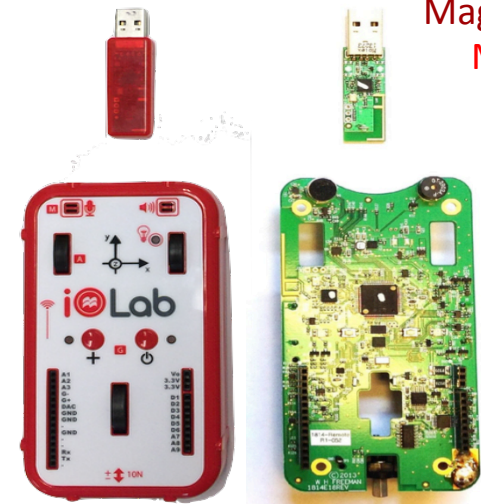
Eugenia Etkina et. al. *The Journal of the Learning Sciences*, 19: 54–98, (2010)

Students benefit more from activities that stress creativity, communication, and sense-making.

Collaborators



www.IOLab.science



- Magnetometer
- Microphone
- Buzzer
- Light Intensity
- Accelerometer
- Gyroscope
- Temperature
- Pressure
- Coded wheel
- Force Probe
- Expansion Headers

Our New Approach:

Each student has their own
(powerful, simple) lab system



Tools

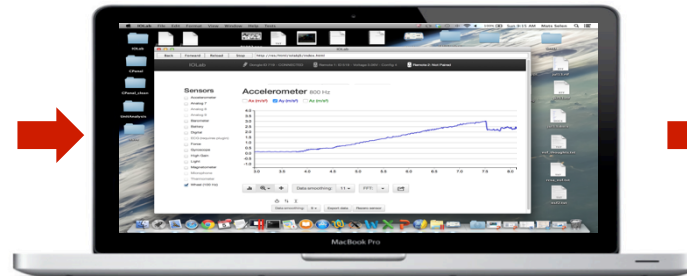
Students explore individual lab
activities at home, followed up
by group activities in class



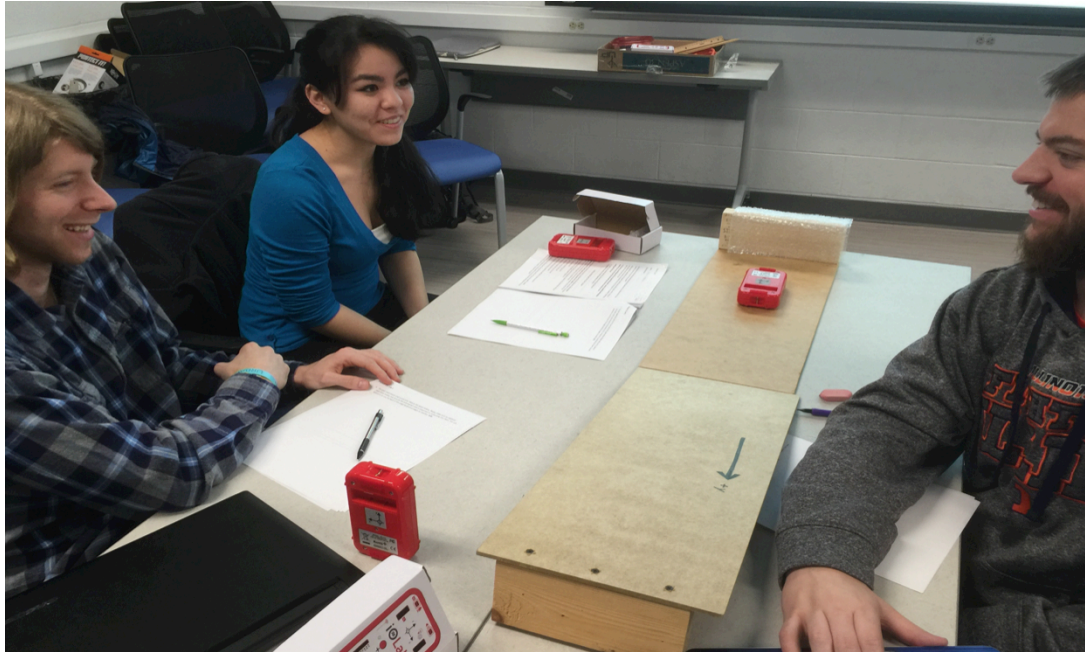
Time



Katie Ansell



Students answer
questions and
share data online



Students are challenged to:

- design experiments
- make sense of data
- work together
- work harder
- work longer

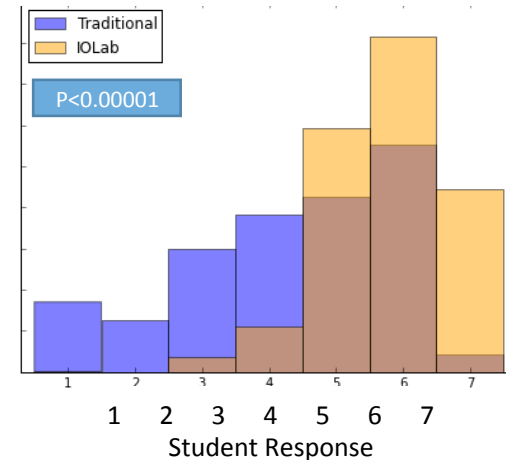
Students like new approach...

“I'm used to the dry cut, fill in the blanks crap that I was subjected to in high school. This physics labs challenges me to think for myself and be curious. I really appreciate that.”



Students want to be challenged !

Labs fun and interesting



Summary

- What is most important for learning ?
 - Student engagement with topic
 - Student engagement with instructor during precious class-time
- What are the challenges to delivering this ?
 - High enrolment (must be scalable)
 - Limited budgets (both students and institutions)
 - Not all technology is pedagogically good (i.e. distraction)
 - *Faculty confidence, buy-in, and training*
- What are the opportunities are available to help ?
 - Scalable technology validated by discipline based education research